

$$(a) \frac{\partial C}{\partial t} = \tilde{D} \frac{\partial^2 C}{\partial x^2}$$

$$\frac{\partial C}{\partial t} = R(\beta) (12 \text{ at} \% \text{ Au}) \exp[R(\beta)t] \cos \beta x$$

$$\frac{\partial^2 C}{\partial x^2} = -\beta^2 (12 \text{ at} \% \text{ Au}) \exp[R(\beta)t] \cos \beta x$$

$$\therefore R(\beta) (12 \text{ at} \% \text{ Au}) \exp[R(\beta)t] \cos \beta x$$

$$= -\tilde{D} \beta^2 (12 \text{ at} \% \text{ Au}) \exp[R(\beta)t] \cos \beta(x) \Rightarrow R(\beta) = -\tilde{D} \beta^2$$

$$C_{\max} = (38 \text{ at} \% \text{ Au}) + (12 \text{ at} \% \text{ Au}) \exp[R(\beta)t] \quad @ \beta x = n\pi \text{ \& n is even} //$$

$$C_{\min} = (38 \text{ at} \% \text{ Au}) - (12 \text{ at} \% \text{ Au}) \exp[R(\beta)t] \quad @ \beta x = n\pi \text{ \& n is odd}$$

$$C_{\max} - C_{\min} = 2(12 \text{ at} \% \text{ Au}) \exp[R(\beta)t] = (24 \text{ at} \% \text{ Au})$$

$$12 \exp[R(\beta)t] = 1 \Rightarrow 12 \exp[-\tilde{D} \beta^2 t] = 1$$

$$t = \frac{\ln(1/12)}{\tilde{D} \beta^2} = \frac{\ln(1/12)}{(10^{-23} \text{ m}^2/\text{s}) \left(\frac{\pi}{10^{-9} \text{ m}}\right)^2} \approx 2.977 \text{ s}$$

$$(b) \frac{\partial C}{\partial t} = \tilde{D} \frac{\partial^2 C}{\partial x^2} - \frac{2k\tilde{D}}{f''} \frac{\partial^4 C}{\partial x^4}$$

$$\frac{\partial^4 C}{\partial x^4} = \beta^4 (12 \text{ at} \% \text{ Au}) \exp[R(\beta)t] \cos \beta x$$

$$\Rightarrow R(\beta) (12 \text{ at} \% \text{ Au}) \exp[R(\beta)t] \cos \beta x$$

$$= -\tilde{D} \beta^2 (12 \text{ at} \% \text{ Au}) \exp[R(\beta)t] \cos \beta x$$

$$- \frac{2k\tilde{D}}{f''} \beta^4 (12 \text{ at} \% \text{ Au}) \exp[R(\beta)t] \cos \beta x$$

$$R(\beta) = -\tilde{D} \beta^2 - \frac{2k\tilde{D}}{f''} \beta^4$$

$$C_{max} - C_{min} = 2(12 \text{ at } \% \text{ Au}) \exp[R(\beta)t] = 2 \text{ at } \% \text{ Au}$$

$$12 \exp[R(\beta)t] = 12 \exp\left[\left(\bar{D}\beta^2 - \frac{2k\bar{D}}{f''} \beta^4\right)t\right] = 1$$

$$t = \frac{\ln(1/2)}{\bar{D}\beta^2 - \frac{2k\bar{D}}{f''} \beta^4} = \frac{\ln(1/2)}{(10^{-23} \text{ m}^2/\text{s})\left(\frac{\pi}{10^{-9} \text{ m}}\right)^2 + \frac{2(-2.16 \times 10^{-11} \text{ J/m})(10^{-23} \text{ m}^2/\text{s})\left(\frac{\pi}{10^{-9} \text{ m}}\right)^4}{(5 \times 10^9 \text{ J/m}^3)}}$$

$$\approx 28057 \text{ s}$$

(c) Ag-Au system: bond \approx $\frac{1}{2}$ negative gradient energy coefficient \approx $\frac{1}{2}$.

계면 에너지를 고려한다면, spinodal decomposition을 유도할 수 있다.

(b)의 경우와 같다.