

1.

a) liquid NaF의 증기압곡선 1 atm이 될 때의 온도를 구하라.

$$(0 \Rightarrow) \ln 1 = -\frac{31090}{T} - 2.52 \ln T + 34.66, T = 2006 \text{ K}$$

b) Triple point 의 온도를 T_T , 압력을 P_T 라고 하면

$$\ln P_T = -\frac{34450}{T_T} - 2.01 \ln T_T + 33.74$$

$$\ln P_T = -\frac{31090}{T_T} - 2.52 \ln T_T + 34.66$$

$$\text{위의 연립 방정식을 풀면, } T_T = 1239 \text{ K}, P_T = e^{-\frac{34450}{1239} - 2.01 \ln 1239 + 33.74} = 2.287 \times 10^{-4} \text{ atm}$$

c) Clausius-Clapeyron equation.

$$\left(\frac{dP}{dT}\right)_{eq} = \frac{\Delta H}{RT^2} \Rightarrow \frac{1}{P} \left(\frac{dP}{dT}\right)_{eq} = \frac{\Delta H}{RT^2} \rightarrow \frac{d}{dT} \ln P = \frac{\Delta H}{RT^2}$$

$$\frac{d}{dT} \ln P = \frac{d}{dT} \left(-\frac{31090}{T} - 2.52 \ln T + 34.66 \right) = \frac{31090}{T^2} - \frac{2.52}{T} = \frac{\Delta H}{RT^2}$$

$$\rightarrow \Delta H = 31090R - 2.52RT = R(31090 - 2.52T)$$

T에 위에서 구한 normal boiling temperature를 대입하면,

$$\Delta H = 8.3145(31090 - 2.52 \times 2006) = 216.5 \text{ kJ/mol}$$

d) ~~the~~ melting ΔH 를 구해야 하는 데 주어진 식이 없으므로

$$\Delta H_{s \rightarrow l} = \Delta H_{s \rightarrow g} + \Delta H_{g \rightarrow l} = \Delta H_{s \rightarrow g} - \Delta H_{l \rightarrow g} \text{로 구한다.}$$

$$\Delta H_{l \rightarrow g} = R(31090 - 2.52T) \text{ (c)에서 구함)} \dots \text{①}$$

$$\text{c)의 방법을 } \Delta H_{s \rightarrow g} \text{를 구해보면, } \frac{d}{dT} \ln P = \frac{34450}{T^2} - \frac{2.01}{T} = \frac{\Delta H_{s \rightarrow g}}{RT^2}$$

$$\Delta H_{s \rightarrow g} = R(34450 - 2.01T) \dots \text{②}$$

$$\Delta H_{s \rightarrow l} = \text{②} - \text{①} = R(3360 + 0.51T), T = T_T = 1239 \text{ K 대입하면}$$

$$\Delta H_{s \rightarrow l} = 33.19 \text{ kJ/mol}$$

e) solid, $T_0 \xrightarrow{\text{// } T_m}$ solid, $T_{\text{melting point}} \rightarrow$ liquid, $T_{\text{melting point}} \rightarrow$ liquid, T_0

$$\Delta H_{s \rightarrow l} = C_{p, \text{solid}}(T_m - T_0) + \Delta H_{\text{melting}} + C_{p, \text{liquid}}(T_0 - T_m)$$

$$= \text{some polynomial} + (C_{p, \text{liquid}} - C_{p, \text{solid}})T$$

$$\Delta H_{s \rightarrow l} = R(3360 + 0.51T) \text{ 에서 } R(0.51) = C_{p, \text{liquid}} - C_{p, \text{solid}} \text{ 라고 할 수 있고 그 값은 } 4.24 \text{ J/mol} \cdot \text{K}$$

2.

$$\text{bulk 상태의 } \Delta G_{\text{bulk}} = \Delta H - T\Delta S$$

$$\text{nanoparticle 상태의 } \Delta G_{\text{n.p.}} = \Delta H - T\Delta S + V_m \left(\frac{2\gamma_s}{r} - \frac{2\gamma_e}{r} \right)$$

각 상태의 melting point를 ~~T_m~~ $T_{m,b}$, $T_{m,n}$ 으로 표현

$$\Delta G_{\text{bulk},m} = \Delta H_m - T_{m,b} \Delta S_m = 0$$

$$\Delta G_{\text{n.p.},m} = \Delta H_m - T_{m,n} \Delta S_m + V_m \left(\frac{2\gamma_s}{r} - \frac{2\gamma_e}{r} \right) = 0$$

$$\Delta G_{\text{bulk},m} = \Delta G_{\text{n.p.},m}$$

$$\Delta H_m - T_{m,b} \Delta S_m = \Delta H_m - T_{m,n} \Delta S_m + V_m \left(\frac{2\gamma_s}{r} - \frac{2\gamma_e}{r} \right)$$

$$(T_{m,n} - T_{m,b}) \Delta S_m = V_m \left(\frac{2\gamma_s}{r} - \frac{2\gamma_e}{r} \right)$$

$$\text{melting point difference} = \frac{2}{r} V_m \frac{\gamma_s - \gamma_e}{\Delta S_m}$$

$$\frac{\Delta T_m}{T} = \frac{2}{r} V_m \frac{\gamma_s - \gamma_e}{T_m \Delta S_m}$$

$$= \frac{2}{r} V_m \frac{\gamma_s - \gamma_e}{\Delta H_m}$$



3.

$$Q_2 = Q + (1-x_2) \frac{dQ}{dx_2} \bigg|_{x_2=2}$$

$$\bar{G}_A = G_m + (1-x_A) \frac{dG_m}{dx_A} = G_m + (1-x_A) [\bar{G}_A - \bar{G}_B + RT \{ \ln x_A - \ln(1-x_A) \} + (1-2x_A) \{ L_0 + L_1(2x_A-1) \} + x_A(1-x_A) 2L_1]$$

$$= x_A \bar{G}_A + x_B \bar{G}_B + RT \{ x_A \ln x_A + x_B \ln x_B \} + x_A x_B \{ L_0 + (x_A - x_B) L_1 \} +$$

$$= \bar{G}_A + RT \ln x_A + x_B^2 L_0 + x_B^2 (4x_A - 1) L_1$$

$$x_A^2 (\bar{G}_A - \bar{G}_B) = x_A^2 L_1$$

$$\bar{G}_B = G_m + (1-x_B) \frac{dG_m}{dx_B} = \bar{G}_B + RT \ln x_B + x_A^2 L_0 + x_A^2 (1-4x_B) L_1$$

$$b) RT \ln a_A = \bar{G}_A - \bar{G}_A = RT \ln x_A + x_B^2 L_0 + x_B^2 (4x_A - 1) L_1$$

$$i) x_A \rightarrow 1, RT \ln a_A \rightarrow 0, a_A \rightarrow 1$$

a_A 가 1에 근사한다는 것은 ideal gas model이다. Raoult's law

$$ii) x_A \rightarrow 0, RT \ln a_A = RT \ln x_A + L_0 - L_1,$$

$$a_A = e^{x_A + \frac{L_0 - L_1}{RT}} \rightarrow e^{\frac{L_0 - L_1}{RT}} \text{ (constant)} \quad \text{Henry's law}$$

$$c) G^m = \sum x_i \bar{G}_i = x_A \bar{G}_A + x_B \bar{G}_B$$

$$\frac{x_B \bar{G}_B}{x_B} = x_A \bar{G}_A + x_B \bar{G}_B + RT \{ x_A \ln x_A + x_B \ln x_B \} + x_A x_B \{ L_0 + (x_A - x_B) L_1 \}$$

$$= x_A \bar{G}_A - x_A RT \ln x_A - x_A x_B^2 L_0 + x_A x_B^2 (4x_A - 1) L_1$$

$$= x_B \bar{G}_B + RT x_B \ln x_B + x_A x_B (1 - x_B) L_0 + x_A x_B (x_A - x_B - 4x_A x_B + x_B) L_1$$

$$= x_B \bar{G}_B + RT x_B \ln x_B + x_A x_B (1 - x_B) L_0 + x_A x_B (1 - 4x_B) L_1$$

$$\bar{G}_B = \bar{G}_B + RT \ln x_B + x_A (1 - x_B) L_0 + x_A^2 (1 - 4x_B) L_1$$

4.

Regular Solution

$$G_m = X_A G_A^\circ + X_B G_B^\circ + RT(X_A \ln X_A + X_B \ln X_B) + X_A X_B L_{AB}$$

liquid에 대하여, $X_A G_A^{\text{ref} \rightarrow l} + X_B G_B^{\text{ref} \rightarrow l} + RT(X_A \ln X_A + X_B \ln X_B) + X_A X_B L_{AB}(l) =: G_m(l)$

solid에 대하여, $X_A G_A^{\text{ref} \rightarrow s} + X_B G_B^{\text{ref} \rightarrow s} + RT(X_A \ln X_A + X_B \ln X_B) + X_A X_B L_{AB}(s) =: G_m(s)$

$$X_A = 1 - X_B \text{ 대입}$$

$$G_m(l) = (1 - X_B) G_A^{\text{ref} \rightarrow l} + X_B G_B^{\text{ref} \rightarrow l} + RT((1 - X_B) \ln(1 - X_B) + X_B \ln X_B) + (1 - X_B) X_B L_{AB}(l)$$

$$G_m(s) = (1 - X_B) G_A^{\text{ref} \rightarrow s} + X_B G_B^{\text{ref} \rightarrow s} + RT((1 - X_B) \ln(1 - X_B) + X_B \ln X_B) + (1 - X_B) X_B L_{AB}(s)$$

 X_B 에 대해 미분

$$\frac{\partial G_m(l)}{\partial X_B} = -G_A^{\text{ref} \rightarrow l} + G_B^{\text{ref} \rightarrow l} + RT(-\ln(1 - X_B) + \ln X_B) + (1 - 2X_B) L_{AB}(l)$$

$$\frac{\partial G_m(s)}{\partial X_B} = -G_A^{\text{ref} \rightarrow s} + G_B^{\text{ref} \rightarrow s} + RT(-\ln(1 - X_B) + \ln X_B) + (1 - 2X_B) L_{AB}(s)$$

 X_B 에 $X_B(s)$ 를 대입한 값 = $X_B(l)$ 을 대입한 값.

$$-G_A^{\text{ref} \rightarrow l} + G_B^{\text{ref} \rightarrow l} + G_A^{\text{ref} \rightarrow s} - G_B^{\text{ref} \rightarrow s} + RT\left(\ln \frac{1 - X_B(s)}{1 - X_B(l)} + \ln \frac{X_B(l)}{X_B(s)}\right) + (1 - 2X_B(l)) L_{AB}(l) - (1 - 2X_B(s)) L_{AB}(s) = 0$$

$$\text{I}_2 \text{에 } \Delta G_m(A) = G_A^{\text{ref} \rightarrow l} - G_A^{\text{ref} \rightarrow s}, \Delta G_m(B) = G_B^{\text{ref} \rightarrow l} - G_B^{\text{ref} \rightarrow s} \text{ 이므로 위의 식을 다시 쓰면 } =: \text{remainder}$$

$$\textcircled{\times} -\Delta G_m(A) + \Delta G_m(B) + \text{remainder} = 0. \text{ 이따 } \Delta G_m(A), \Delta G_m(B) \text{ 는 constant 이고}$$

remainder에 는 reference term이 없으므로 unique 값을 증명할 수 있다.

5.

model

 $a_i \neq X_i$ 이므로 ideal solution이 아니다

regular solution model에서

$$G_m = X_A G_A^\circ + X_B G_B^\circ + RT(X_A \ln a_A + X_B \ln a_B) = X_A G_A^\circ + X_B G_B^\circ + RT(X_A \ln X_A + X_B \ln X_B) + RT(X_A \ln \gamma_A + X_B \ln \gamma_B)$$

$$\text{이므로 } X_A X_B L_{AB} = RT(X_A \ln \gamma_A + X_B \ln \gamma_B) \quad \left(\gamma_A = \frac{a_A}{X_A}, \gamma_B = \frac{a_B}{X_B}\right)$$

$$RT \ln \gamma_A = L_{AB}$$

900K

