

1.

a) liquid NaF의 증기압의 1 atm이 될 때의 온도를 구한다.

$$(0 =) \ln 1 = -\frac{31090}{T} - 2.52 \ln T + 34.66, T = 2006K$$

b) Triple point의 온도를 T_T , 압력을 P_T 라고 하면

$$\ln P_T = -\frac{34450}{T_T} - 2.01 \ln T_T + 33.74$$

$$\ln P_T = -\frac{31090}{T_T} - 2.52 \ln T_T + 34.66$$

$$\text{위의 연립방정식을 풀면, } \boxed{T_T = 1239K}, P_T = e^{-\frac{34450}{1239} - 2.01 \ln 1239 + 33.74} = 2.287 \times 10^4 \text{ atm}$$

c) Clausius-Clapeyron equation.

$$\left(\frac{dP}{dT}\right)_{eq} = \frac{\Delta H}{RT^2} \Rightarrow \frac{1}{P} \left(\frac{dP}{dT}\right)_{eq} = \frac{\Delta H}{RT^2} \Rightarrow \frac{d}{dT} \ln P = \frac{\Delta H}{RT^2}$$

$$\frac{d}{dT} \ln P = \frac{d}{dT} \left(-\frac{31090}{T} - 2.52 \ln T + 34.66 \right) = \frac{31090}{T^2} - \frac{2.52}{T} = \frac{\Delta H}{RT^2}$$

$$\rightarrow \Delta H = 31090R - 2.52RT = R(31090 - 2.52T)$$

T 에 뷔에서 구한 normal boiling temperature를 대입하면,

$$\Delta H = 8.345(31090 - 2.52 \times 2006) = 216.5 \text{ kJ/mol.}$$

d) melting ΔH 를 구해야 하는데 주어진 식이 없으므로

$$\Delta H_{s \rightarrow l} = \Delta H_{s \rightarrow g} + \Delta H_{g \rightarrow l} = \Delta H_{s \rightarrow g} - \Delta H_{l \rightarrow g} \text{로 구한다.}$$

$$\Delta H_{l \rightarrow g} = R(31090 - 2.52T) \quad (\text{c)에서 구함}) \quad \dots \quad ①$$

$$\text{c)의 방법으로 } \Delta H_{s \rightarrow g} \text{를 구해보면, } \frac{d}{dT} \ln P = \frac{34450}{T^2} - \frac{2.01}{T} = \frac{\Delta H_{s \rightarrow g}}{RT^2}$$

$$\Delta H_{s \rightarrow g} = R(34450 - 2.01T) \quad \dots \quad ②$$

$$\Delta H_{s \rightarrow l} = ② - ① = R(3360 + 0.51T), T = T_T = 1239K \text{ 대입하면}$$

$$\Delta H_{s \rightarrow l} = 33.19 \text{ kJ/mol.}$$

// T_m

e) solid, $T_0 \rightarrow$ solid, $T_{\text{melting point}} \rightarrow$ liquid, $T_{\text{melting point}} \rightarrow$ liquid, T_0

$$\Delta H_{s \rightarrow l} = C_{p,\text{solid}}(T_m - T_0) + \Delta H_{\text{melting}} + C_{p,\text{liquid}}(T_0 - T_m)$$

$$= \cancel{C_p} \text{ (some polynomial)} + (C_{p,\text{liquid}} - C_{p,\text{solid}})T$$

$$\Delta H_{s \rightarrow l} = R(3360 + 0.51T) \text{에서 } \cancel{R(0.51)} = C_{p,\text{liquid}} - C_{p,\text{solid}} \text{라고 할 수 있고 그 값은 } 4.24 \text{ J/mol.K}$$

2.

$$\text{bulk상태의 } \Delta G_{\text{bulk}} = \Delta H - T\Delta S$$

$$\text{nanoparticle 상태 } \Delta G_{n,p} = \Delta H - T\Delta S + V_m \left(\frac{2\gamma_s}{r} - \frac{2\gamma_e}{r} \right)$$

각상의 melting point를 ~~T_{m,f}~~ T_{m,b}, T_{m,n}으로 두면

$$\Delta G_{\text{bulk},m} = \Delta H_m - T_{m,f} \Delta S_m = 0$$

$$\Delta G_{n,p,m} = \Delta H_m - T_{m,n} \Delta S_m + V_m \left(\frac{2\gamma_e}{r} - \frac{2\gamma_s}{r} \right) = 0$$

$$\Delta G_{\text{bulk},m} = \Delta G_{n,p,m}$$

$$\Delta H_m - T_{m,b} \Delta S_m = \Delta H_m - T_{m,n} \Delta S_m + V_m \left(\frac{2\gamma_e}{r} - \frac{2\gamma_s}{r} \right)$$

$$(T_{m,n} - T_{m,b}) \Delta S_m = V_m \left(\frac{2\gamma_e}{r} - \frac{2\gamma_s}{r} \right)$$

$$\textcircled{\Phi} \text{ melting point 증감도} = \frac{2}{r} V_m \cdot \frac{\gamma_s - \gamma_e}{\Delta S_m}$$

$$\frac{\Delta T_m}{T} = \frac{2}{r} V_m \frac{\gamma_s - \gamma_e}{T_m \Delta S_m}$$

$$= \frac{2}{r} V_m \frac{\gamma_s - \gamma_e}{\Delta H_m}$$

(B)

3

$$Q_2 = Q + (1-x_2) \frac{dQ}{dx_2} \Big|_{x_2=2}$$

~~(*)~~ $\bar{G}_A = G_m + (1-x_A) \frac{dG_m}{dx_A} = G_m + (1-x_A) \left[{}^0G_A - {}^0G_B + RT \{ \ln x_A - \ln (1-x_A) \} + (1-2x_A) \{ L_o + L_i(2x_A-1) \} + x_A(1-x_A)2L_i \right]$

$$= {}^0G_A + x_A {}^0G_B + RT \{ x_A \ln x_A + x_B \ln x_B + x_A x_B \{ L_o + (x_A - x_B)L_i \} \} +$$

$$= {}^0G_A + RT \ln x_A + x_A^2 L_o + x_B^2 (4x_A - 1) L_i$$

$$\bar{G}_B = G_m + (1-x_B) \frac{dG_m}{dx_B} = {}^0G_B + RT \ln x_B + x_A^2 L_o + x_B^2 (4x_A - 1) L_i$$

b) $RT \ln \alpha_A = \bar{G}_A - {}^0G_A = RT \ln x_A + x_B^2 L_o + x_B^2 (4x_A - 1) L_i$

i) $x_A \rightarrow 1, RT \ln \alpha_A \rightarrow 0, \alpha_A \rightarrow 1$

$\alpha_A \approx 1$ 에 근사한다는 것은 ideal 한 model이다. [Raoult's law]

ii) $x_A \rightarrow 0, RT \ln \alpha_A = RT \ln x_A + L_o - L_i$,

$$\alpha_A = e^{x_A + \frac{L_o - L_i}{RT}} \rightarrow e^{\frac{L_o - L_i}{RT}} \text{ (constant)} \quad [\text{Henry's law}]$$

c) $G^m = \sum x_i \bar{G}_i = x_A \bar{G}_A + x_B \bar{G}_B$

$$\begin{aligned} x_B \bar{G}_B &= x_A {}^0G_A + x_B {}^0G_B + RT \{ x_A \ln x_A + x_B \ln x_B \} + x_A x_B \{ L_o + (x_A - x_B)L_i \} \\ &= x_A {}^0G_A - x_A RT \ln x_A - x_A x_B^2 L_o + x_A x_B^2 (4x_A - 1) L_i \end{aligned}$$

$$= x_B {}^0G_B + RT x_B \ln x_B + x_A x_B (1-x_B) L_o + x_A x_B (x_A - x_B - 4x_A x_B + x_B) L_i$$

$$= x_B {}^0G_B + RT x_B \ln x_B + x_A x_B (1-x_B) L_o + x_A x_B (1-4x_B) L_i$$

$$\bar{G}_B = {}^0G_B + RT \ln x_B + x_A (1-x_B) L_o + x_A^2 (1-4x_B) L_i$$

4.

Regular Solution

$$G_m = X_A G_A^\circ + X_B G_B^\circ + RT(X_A \ln X_A + X_B \ln X_B) + X_A X_B L_{AB}$$

liquid에 대하여, $X_A G_A^{\text{ref}\rightarrow l} + X_B G_B^{\text{ref}\rightarrow l} + RT(X_A \ln X_A + X_B \ln X_B) + X_A X_B L_{AB}(l) =: G_m(l)$

solid에 대하여, $X_A G_A^{\text{ref}\rightarrow s} + X_B G_B^{\text{ref}\rightarrow s} + RT(X_A \ln X_A + X_B \ln X_B) + X_A X_B L_{AB}(s) =: G_m(s)$

$$X_A = 1 - X_B \text{ 대입}$$

$$G_m(l) = (1-X_B) G_A^{\text{ref}\rightarrow l} + X_B G_B^{\text{ref}\rightarrow l} + RT((1-X_B) \ln(1-X_B) + X_B \ln X_B) + (1-X_B) X_B L_{AB}(l)$$

$$G_m(s) = (1-X_B) G_A^{\text{ref}\rightarrow s} + X_B G_B^{\text{ref}\rightarrow s} + RT((1-X_B) \ln(1-X_B) + X_B \ln X_B) + (1-X_B) X_B L_{AB}(s)$$

X_B 에 대해 미분

$$\frac{\partial G_m(l)}{\partial X_B} = -G_A^{\text{ref}\rightarrow l} + G_B^{\text{ref}\rightarrow l} + RT(-\ln(1-X_B) + \ln X_B) + (1-2X_B) L_{AB}(l)$$

$$\frac{\partial G_m(s)}{\partial X_B} = -G_A^{\text{ref}\rightarrow s} + G_B^{\text{ref}\rightarrow s} + RT(-\ln(1-X_B) + \ln X_B) + (1-2X_B) L_{AB}(s)$$

X_B 에 $X_B(s)$ 를 대입한 값 = $X_B(l)$ 을 대입한 값

$$-G_A^{\text{ref}\rightarrow l} + G_B^{\text{ref}\rightarrow l} + G_A^{\text{ref}\rightarrow s} - G_B^{\text{ref}\rightarrow s} + RT\left(\ln \frac{1-X_B(s)}{1-X_B(l)} + \ln \frac{X_B(l)}{X_B(s)}\right) + (1-2X_B(l)) L_{AB}(l) - (1-2X_B(s)) L_{AB}(s) = 0$$

그리하여 $\Delta G_m(A) = G_A^{\text{ref}\rightarrow l} - G_A^{\text{ref}\rightarrow s}$, $\Delta G_m(B) = G_B^{\text{ref}\rightarrow l} - G_B^{\text{ref}\rightarrow s}$ 이므로 위의식을 다시쓰면 $=: \text{remainder}$

$\Delta G_m(A) + \Delta G_m(B) + \text{remainder} = 0$. 이때 $\Delta G_m(A)$, $\Delta G_m(B)$ 는 constant이고 remainder는 reference term이었으므로 unique함을 증명할 수 있다.

5.

model

$\alpha_A \neq X_A$ 이므로 ideal solution이 아님

regular solution model에 대하여

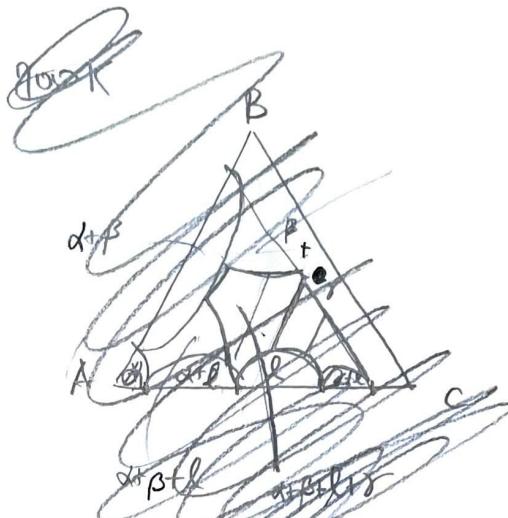
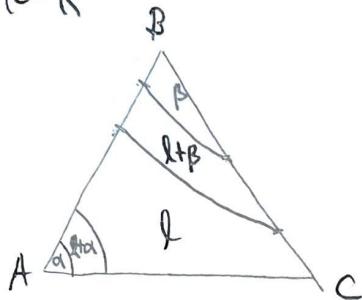
$$G_m = X_A G_A^\circ + X_B G_B^\circ + RT(X_A \ln \gamma_A + X_B \ln \gamma_B) = X_A G_A^\circ + X_B G_B^\circ + RT(X_A \ln X_A + X_B \ln X_B) + RT(X_A \ln \gamma_A + X_B \ln \gamma_B)$$

이므로 $X_A X_B L_{AB} = RT(X_A \ln \gamma_A + X_B \ln \gamma_B)$ ($\gamma_A = \frac{\alpha_A}{X_A}$, $\gamma_B = \frac{\alpha_B}{X_B}$)

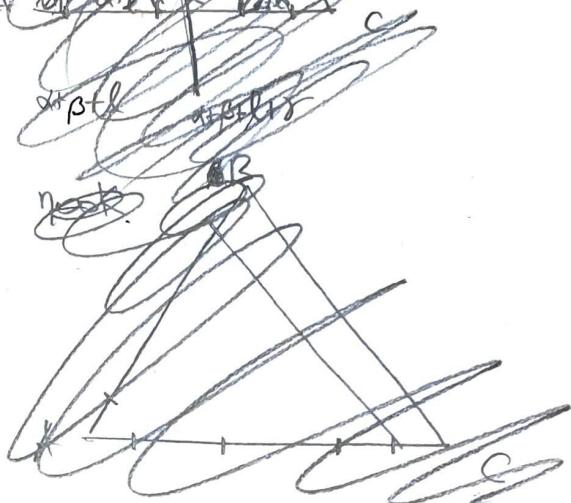
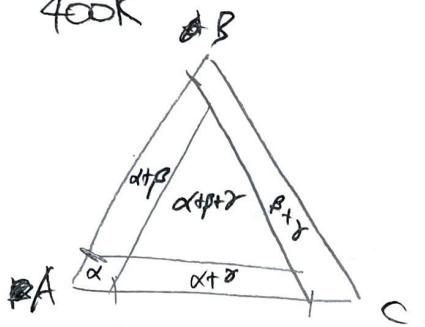
$$RT \ln \gamma_A = L_{AB}$$

6.

900K



400K



1100K)

