

REPORT



POSTECH
POHANG UNIVERSITY OF SCIENCE AND TECHNOLOGY

제목 : **Homework #5**

수강과목 : **상변태론**

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상변태론 HW #5

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- a) For a spherical nucleus, derive the following expression, the energy change during nucleation as a function of number of atoms n in cluster (v is atomic volume).

$$\Delta G = -n \Delta G_a + (36\pi)^{\frac{1}{3}} n^{\frac{2}{3}} v^{\frac{2}{3}} \gamma$$

Sol) $V_{\text{total}} = n (\text{number of atom}) \times V (\text{atomic volume})$

$$= \frac{4}{3} \pi r^3 = nV \quad \text{--- ①}$$

$$A_{\text{total}} = 4\pi r^2 \quad \text{--- ②}$$

②에 ①을 적용하면, $4\pi r^2 = (36\pi)^{\frac{1}{3}} n^{\frac{2}{3}} V^{\frac{2}{3}}$

$$\therefore \Delta G = -nV \Delta G_v + (36\pi)^{\frac{1}{3}} n^{\frac{2}{3}} V^{\frac{2}{3}} \gamma$$

- b) Using the result of (a), derive the expression for the critical number of atoms and energy barrier.

$$\frac{\partial \Delta G}{\partial n} = -V \Delta G_v + (36\pi)^{\frac{1}{3}} V^{\frac{2}{3}} \gamma \times \frac{2}{3} n^{-\frac{1}{3}}, \quad \left. \frac{\partial \Delta G}{\partial n} \right|_{n=n^*} = 0$$

$$V \Delta G_v = (36\pi)^{\frac{1}{3}} V^{\frac{2}{3}} \times \frac{2}{3} n^{*\frac{-1}{3}} \gamma$$

$$n^{*\frac{1}{3}} = \frac{2(36\pi)^{\frac{1}{3}}}{3V \Delta G_v} V^{\frac{2}{3}} \gamma = \frac{2}{3 \Delta G_v} (36\pi)^{\frac{1}{3}} \times \frac{1}{V^{\frac{1}{3}}} \gamma$$

$$\therefore n^* = \frac{8 \times 36 \pi^{\frac{4}{3}}}{32\pi \Delta G_v^3 \times V} \gamma^3 = \frac{32\pi}{3V \Delta G_v^3} \gamma^3$$

$$\therefore \Delta G^* = -\frac{32\pi \gamma^3}{3V \Delta G_v^3} V \Delta G_v + (36\pi)^{\frac{1}{3}} \left(\frac{32\pi}{3} \right)^{\frac{2}{3}} \left(\frac{\gamma^2}{\Delta G_v^2} \right) \gamma = \frac{16\pi}{3} \frac{\gamma^3}{\Delta G_v^2}$$

- c) Assuming isotropic and constant surface energy for both of graphite and diamond,

and using the data : $\gamma_{\text{gr}} = 3.1 \text{ Jm}^{-2}$, $\gamma_{\text{dia}} = 3.6, 3.65 \text{ and } 3.7 \text{ Jm}^{-2}$, respectively

$\gamma_{\text{gr}} = 8 \text{ \AA}^3/\text{atom}$, $\gamma_{\text{dia}} = 6 \text{ \AA}^3/\text{atom}$, ${}^{\circ}\text{G}_{\text{dia}} - {}^{\circ}\text{G}_{\text{gr}} = 0.02 \text{ eV/atom}$

- a) 결과를 참고하면, $\Delta G_s = -n ({}^{\circ}\text{G}_{\text{dia}} - {}^{\circ}\text{G}_{\text{gr}}) V + (36\pi)^{\frac{1}{3}} n^{\frac{2}{3}} (r_{\text{dia}} \cdot V_{\text{dia}} - r_{\text{gr}} \cdot V_{\text{gr}})$ 이고

isotropic & constant 일 경우에서는 $\Delta G_s = 0$, $n = 36\pi \left(\frac{V_{\text{dia}}^{\frac{2}{3}} \cdot r_{\text{dia}} - V_{\text{gr}}^{\frac{2}{3}} \cdot r_{\text{gr}}}{{}^{\circ}\text{G}_{\text{gr}} - {}^{\circ}\text{G}_{\text{dia}}} \right)^{\frac{3}{2}}$

1) $r_{\text{dia}} = 3.6 \text{ \AA}$

$$n = 36\pi \left(\frac{(6 \times 10^{-30} \text{ m}^3/\text{atom})^{\frac{2}{3}} (3.6 \times \frac{1}{1.6 \times 10^{-19} \text{ eV/m}^2})^{\frac{3}{2}} - (8 \times 10^{-30} \text{ m}^3/\text{atom})^{\frac{2}{3}} (3.1 \times \frac{1}{1.6 \times 10^{-19} \text{ eV/m}^2})^{\frac{3}{2}}}{-0.02 \text{ eV/atom}} \right)$$

$$\approx 465$$

$$2) \gamma_{\text{dia}} = 3.65 \text{ J}$$

$$n = 36\pi \left(\frac{(6 \times 10^{-30} \text{ m}^3/\text{atom})^{2/3} (3.65 \times \frac{1}{1.6 \times 10^{-19}} \text{ eV/m}^2)^3 - (8 \times 10^{-30} \text{ m}^3/\text{atom})^{2/3} (3.1 \times \frac{1}{1.6 \times 10^{-19}} \text{ eV/m}^2)^3}{-0.02 \text{ eV/atom}} \right)$$

$$\approx 146$$

$$3) \gamma_{\text{dia}} = 3.7 \text{ J}$$

$$n = 36\pi \left(\frac{(6 \times 10^{-30} \text{ m}^3/\text{atom})^{2/3} (3.7 \times \frac{1}{1.6 \times 10^{-19}} \text{ eV/m}^2)^3 - (8 \times 10^{-30} \text{ m}^3/\text{atom})^{2/3} (3.1 \times \frac{1}{1.6 \times 10^{-19}} \text{ eV/m}^2)^3}{-0.02 \text{ eV/atom}} \right)$$

$$\approx 21$$

d) What is the necessary condition for a diamond cluster of any size to be more stable than graphite?

$$\Delta G_{\text{dia}} < \Delta G_{\text{gr}}, \Delta G_s = \Delta G_{\text{dia}} - \Delta G_{\text{gr}} < 0$$

$$\therefore n < 36\pi \left(\frac{\frac{V_{\text{dia}}^{2/3} R_{\text{dia}} - V_{\text{gr}}^{2/3} R_{\text{gr}}}{G_{\text{gr}} - G_{\text{dia}}}}{3} \right)^3$$

e) Assuming that the critical number of atoms for graphite nucleation is 100, estimate the driving force for graphite nucleation.

$$n^* = 100, \frac{32\pi \times R_{\text{gr}}^3}{3V_{\text{gr}} \Delta G_n^3} = 100 \text{ (atom)}$$

$$\Delta G_{V,\text{gr}}^3 = \frac{32\pi \times R_{\text{gr}}^3}{300 V_{\text{gr}}}$$

$$\Delta G_{V,\text{gr}} = \left(\frac{32\pi \times (3.1 \text{ J/m}^2)^3}{3 \times (8 \times 10^{-30} \text{ m}^3/\text{atom}) \times (100 \text{ atom})} \right)^{1/3} \approx 1.08 \times 10^{10} \text{ J/m}^3$$

f) For the three values of surface energy of diamond, compute the ratio of nucleation rate between graphite and diamond, $I_{\text{gra}}/I_{\text{dia}}$.

For the nucleation rate, use the expression: $I = A \cdot \exp(-\Delta G^*/kT)$, and assume that A is the same constant for both of graphite and diamond and T = 300 K.

$$\frac{I_{\text{gr}}}{I_{\text{dia}}} \text{ 은 주어진 nucleation rate 식에 따랐을 때 } \exp\left(\frac{\Delta G_{\text{dia}}^* - \Delta G_{\text{gr}}^*}{kT}\right)$$

$$\Delta G_{\text{gr}}^* = \frac{16\pi R_{\text{gr}}^3}{3 \Delta G_{V,\text{gr}}^2}, n^* = 100 \text{ 이라고 하고 (e) 결과를 활용하면}$$

$$= \frac{16\pi (3.1 \text{ J/m}^2)^3}{3 (1.08 \times 10^{10} \text{ J/m}^3)^2} \approx 4.28 \times 10^{-10} \text{ J}$$

$$\Delta G_{\text{dia}}^* = \frac{16\pi \gamma_{\text{dia}}^3}{3 \underbrace{\Delta G_{\text{v.dia}}^2}_{\text{를 구해야 함}}}$$

$$\begin{aligned}\Delta G_{\text{v.dia}} \times V_{\text{dia}} &= {}^\circ G_{\text{v}} - {}^\circ G_{\text{dia}} \\ &= \Delta G_{\text{v.gr}} V_{\text{gr}} + {}^\circ G_{\text{gr}} - {}^\circ G_{\text{dia}} \quad (\Delta G_{\text{v.gr}} \cdot V_{\text{gr}} = {}^\circ G_{\text{v}} - {}^\circ G_{\text{gr}}) \\ &= \Delta G_{\text{v.gr}} V_{\text{gr}} - 0.02 \text{ eV}\end{aligned}$$

$n^* = 100$ 이라고 하면,

$$\begin{aligned}\Delta G_{\text{v.dia}} &= \frac{(1.08 \times 10^{10} \text{ J/m}^3) \times (8 \times 10^{-30} \text{ m}^3/\text{atom}) - 0.02 \times 1.6 \times 10^{-19} \text{ J/atom}}{6 \times 10^{-30} \text{ m}^3/\text{atom}} \\ &\simeq 1.39 \times 10^{10} \text{ J/m}^3\end{aligned}$$

1) $\gamma_{\text{dia}} = 3.6 \text{ J/m}^2$ 일 때,

$$\Delta G_{\text{dia}}^* = \frac{16\pi (3.6)^3}{3 \times (1.39 \times 10^{10})^2} \text{ J} = 4.04 \times 10^{-18} \text{ J}$$

$$\begin{aligned}\text{ratio} &= \exp\left(\frac{\Delta G_{\text{dia}}^* - \Delta G_{\text{gr}}^*}{KT}\right) = \exp\left(\frac{(4.04 - 4.28) 10^{-18}}{1.38 \times 10^{-23} \times 300}\right) \\ &= 6.66 \times 10^{-26}\end{aligned}$$

2) $\gamma_{\text{dia}} = 3.65 \text{ J/m}^2$ 일 때,

$$\Delta G_{\text{dia}}^* = \frac{16\pi (3.65)^3}{3 \times (1.39 \times 10^{10})^2} \text{ J} = 4.22 \times 10^{-18} \text{ J}$$

$$\begin{aligned}\text{ratio} &= \exp\left(\frac{\Delta G_{\text{dia}}^* - \Delta G_{\text{gr}}^*}{KT}\right) = \exp\left(\frac{(4.22 - 4.28) 10^{-18}}{1.38 \times 10^{-23} \times 300}\right) \\ &= 5.08 \times 10^{-7}\end{aligned}$$

3) $\gamma_{\text{dia}} = 3.7 \text{ J/m}^2$ 일 때,

$$\Delta G_{\text{dia}}^* = \frac{16\pi (3.7)^3}{3 \times (1.39 \times 10^{10})^2} \text{ J} = 4.39 \times 10^{-18} \text{ J}$$

$$\begin{aligned}\text{ratio} &= \exp\left(\frac{\Delta G_{\text{dia}}^* - \Delta G_{\text{gr}}^*}{KT}\right) = \exp\left(\frac{(4.39 - 4.28) 10^{-18}}{1.38 \times 10^{-23} \times 300}\right) \\ &= 3.46 \times 10^{11}\end{aligned}$$

(g) (c) 결과를 통해서 작은 표면에너지 차이가 입자의 크기를 4~8배 변화시키는 것을 확인할 수 있었고, (f) 결과를 통해서 입자크기가 동일 할 때는 0.05 J/m^2 정도의 surface energy 가 nucleation rate를 약 $10^{17} \sim 10^{18}$ 배의 변화를 이끌어냈다. 두 문제의 결과를 종합하자면, 합성되기 위해 고온, 고압의 공정이 필요한 diamond 도 입자크기가 충분히 작다면 진공 상태의 CVD chamber에서도 합성이 될 수 있다는 예상을 할 수 있다.