

$$pV = nRT \quad (3.1.32)$$

$$G_A = nRT \quad (2.1.52)$$

$$\ln \frac{p}{p^\circ} = - \frac{\Delta G^\circ}{RT} \quad (3.1.32)$$

$$V = NRT^2 \left( \frac{\partial \ln Z}{\partial T} \right)_A \quad (2.1.52) \quad A = \{N, \mu, \dots\}$$

$$Z = \sum e^{-\epsilon_i/kT}$$

$$\epsilon_i = \frac{h^2}{8m} \left( \frac{n_x^2}{a^2} + \frac{n_y^2}{b^2} \right)$$

$$= \sum e^{-\frac{h^2}{8mkT} \left( \frac{n_x^2}{a^2} + \frac{n_y^2}{b^2} \right)} = \sum e^{-\frac{h^2}{8mkT} \frac{n_x^2}{a^2}} \sum e^{-\frac{h^2}{8mkT} \frac{n_y^2}{b^2}}$$

$$= \int_0^\infty e^{-\frac{h^2}{8mkT} \frac{n_x^2}{a^2}} \int_0^\infty e^{-\frac{h^2}{8mkT} \frac{n_y^2}{b^2}}$$

$$= \left( \frac{a}{2} \sqrt{\frac{8\pi mkT}{h^2}} \right) \left( \frac{b}{2} \sqrt{\frac{8\pi mkT}{h^2}} \right)$$

$$= \frac{2ab \pi mkT}{h^2} = \frac{2A \pi mkT}{h^2}$$

$$V = NRT^2 \left( \frac{\partial \ln Z}{\partial T} \right) = NRT^2 \frac{1}{T} = NRT$$

2

$$Z = \sum e^{-\frac{E_i}{kT}}$$

$$N \begin{cases} \text{valency } 0, & E = \Delta H \\ \text{valency } X, & E = \infty \end{cases}$$

$$Z = \sum_{\text{state}} e^{-\frac{E_i}{kT}}$$

$$= e^{-\frac{\Delta H}{kT}} + 1$$

$$\frac{n_v}{N} = \frac{e^{-\frac{\Delta H}{kT}}}{Z} = \frac{e^{-\frac{\Delta H}{kT}}}{1 + e^{-\frac{\Delta H}{kT}}}$$

$$= \frac{1}{1 + e^{\frac{\Delta H}{kT}}}$$

$$\frac{n_v}{N} = \frac{1}{1 + e^{\frac{\Delta H}{kT}}}$$

3

macro state 상태를 나타내, microstate 는 각각 나타낼 수  
 있다. microstate 에서는 각각 가역적인 (reversible) 상태 일 수  
 있지만 이것이 macro state 는 보이면 micro state 일 때의  
 특성이 보이지 않고 비가역적인 (irreversible) 과정 수로  
 나타낼 수 있는 것 같다.

4

$$(a) \quad \Omega = \frac{2N!}{N!N!} \quad \Delta S_{\text{thermal}} = 0$$

$$\Delta S_{\text{config}} = k \ln \Omega$$

$$\begin{aligned} \ln \Omega &= 2N \ln 2N - 2N - (N \ln N - N + N \ln N - N) \\ &= 2N \ln 2N - 2N \ln N \\ &= N \ln 4N^2 - N \ln N^2 = N \ln 4 \end{aligned}$$

$$\Delta S_{\text{config}} = kN \ln 4 = R \ln 4$$

$$\Delta S_{\text{total}} = \Delta S_{\text{thermal}} + \Delta S_{\text{config}} = R \ln 4$$

$$(b) \quad \Delta S_{\text{config}} = \ln \left( \frac{3N!}{2N!N!} \right)$$

$$\begin{aligned} &= 3N \ln 3N - 3N - (2N \ln 2N - 2N + N \ln N - N) \\ &= N \ln 27N^3 - N \ln 4N^2 - N \ln N = N \ln \frac{27}{4} \end{aligned}$$

\* A

$$\Delta S_{\text{thermal}} = 2R \ln \left( \frac{4}{3} \right)$$

\* B

$$\Delta S_{\text{thermal}} = R \ln \left( \frac{2}{3} \right)$$

$$\Delta S_{\text{total}} = \Delta S_{\text{config}} + \Delta S_{\text{thermal}} = R \ln 8$$

(c) 
$$\Delta S_{\text{total}} = 2R \ln 2 + 0 + 3R \ln\left(\frac{2}{3}\right) = R \ln\left(\frac{32}{27}\right)$$