

$$1. a. dU = TdS - PdV \Rightarrow dS = \frac{dU + PdV}{T} = \frac{R}{V}dV + \frac{C_V}{T}dT$$

$$\text{Eq} \Rightarrow dS = \frac{R}{V}dV \quad \Delta S = R \ln \frac{V_2}{V_1}$$

$$V_1 = \frac{nRT}{P_1} = \frac{1 \times 0.08206 \times 300}{10} = 2.46 \quad V_2 = 4.92$$

$$\Delta S = R \ln 2 = 5.763 \text{ J/K}$$

$$b. z=0 \Rightarrow \Delta S=0$$

$$c. \text{Eq} \Rightarrow \Delta S = C_V \ln \frac{T_2}{T_1}$$

$$= 1.5R \ln \frac{1}{2} = -8.6447 \text{ J/K}$$

3.2. a. free expansion $\Rightarrow$ temperature remains constant ($\Delta T = 0$)

$$Q=0, \Delta U=0, \Delta H=0, W=0,$$

ΔS : 상태함수 $\Rightarrow$ 가역 과정 만능 ΔS 를 쓰

$$\hookrightarrow Q = nRT \ln \frac{V_2}{V_1} \rightarrow \Delta S = nR \ln \frac{V_2}{V_1}$$

$$= 1 \times R \ln 3 = 9.1344 \text{ J/K}$$

b. $W=0$

$$Q = \Delta U = C_V (400 - 300) = 1247.17 \text{ J}$$

$$\Delta H = C_P (400 - 300) = 2078.6 \text{ J}$$

$$\Delta S = \int_{300}^{400} \frac{C_V}{T} dT = C_V \ln \frac{4}{3} = 3.58788 \text{ J/K}$$

c. $\Delta U=0, \Delta H=0$

$$Q = W = RT \ln \frac{V_2}{V_1} = R(400) \ln 3 = 3653.75 \text{ J}$$

$$\Delta S = nR \ln 3 = 9.1344 \text{ J/K}$$

d. 쓰

$$\Delta U = C_V (300 - 400) = -1247.17 \text{ J}$$

$$Q = \Delta H = C_P (300 - 400) = -2078.6 \text{ J}$$

$$W = Q - \Delta U = -831.44 \text{ J}$$

$$V_2 = \frac{300}{400} V_1 = \frac{3}{4} V_1$$

$$\Delta S = R \ln \frac{V_2}{V_1} + C_V \ln \frac{T_2}{T_1}$$

$$= R \ln \frac{3}{4} + C_V \ln \frac{3}{4} = -5.9198 \text{ J/K}$$

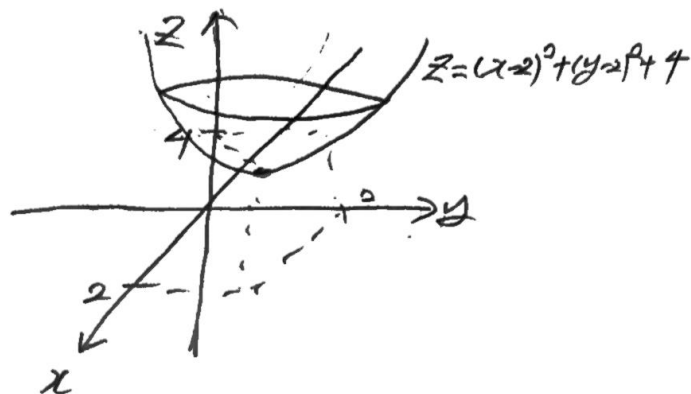
$$5. (a) \frac{\partial Z}{\partial x} = 2(x-2) = 0 \Rightarrow x=2$$

$$\frac{\partial Z}{\partial y} = 2(y-2) = 0 \Rightarrow y=2$$

$$x=2, y=2$$

$$\rightarrow z=4$$

$$\text{at } (2, 2, 4) \Rightarrow \text{minimum value} = 4$$



$$(b) y = 1 - x$$

$$Z = f(x, y) = (x-2)^2 + (y-2)^2 + 4$$

$$\Rightarrow Z = f(x, y) = (x-2)^2 + (-1-x)^2 + 4$$

$$\frac{\partial Z}{\partial x} = 2(x-2) - 2(-1-x)$$

$$= 2x(-4 + 2 + 2x) = 4x - 2 = 0$$

$$x = \frac{1}{2}, y = \frac{1}{2}, Z = (\frac{1}{2}-2)^2 + (\frac{1}{2}-2)^2 + 4 \quad \text{minimum!!}$$

$$= \frac{17}{2} \quad \therefore \frac{17}{2}$$

$$(c) L = (x-2)^2 + (y-2)^2 + 4 + \lambda(x+y-1)$$

$$\frac{\partial L}{\partial x} = 2(x-2) + \lambda = 0 \rightarrow 2x + \lambda = 4$$

$$\frac{\partial L}{\partial y} = 2(y-2) + \lambda = 0 \rightarrow 2y + \lambda = 4 \rightarrow x - y = 0$$

$$\frac{\partial L}{\partial \lambda} = x + y - 1 = 0 \rightarrow x + y = 1$$

$$x = \frac{1}{2}, y = \frac{1}{2}, \lambda = 3$$

$$Z = (\frac{1}{2}-2)^2 + (\frac{1}{2}-2)^2 + 4$$

$$= \frac{17}{2}$$

$$\therefore \frac{17}{2} \quad \text{minimum!!}$$

1. the temperature remain constant $\Rightarrow \Delta U=0$, $Q=W$

$$dS = \frac{PdV}{T} = \frac{nR}{V} dV \Rightarrow \underline{\Delta S = nR \ln\left(\frac{V_2}{V_1}\right)}$$

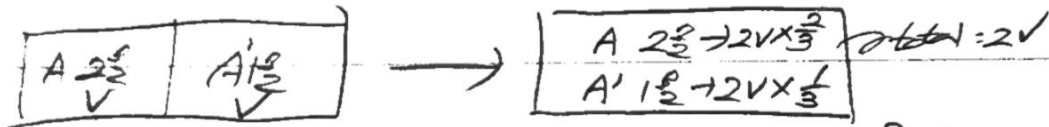
a. $A, B \rightarrow V_2 = 2V_1$

$$\Rightarrow \Delta S = R \ln 2 + R \ln 2 = \underline{2R \ln 2}$$

b. $\Delta S = 2R \ln 2 + R \ln 2 = \underline{3R \ln 2}$

c. $S \rightleftharpoons S$ $A \rightleftharpoons$ equilibrium $\Rightarrow \underline{\Delta S = 0}$

d. $A \rightleftharpoons + A' \rightleftharpoons$



$$\Delta S = \underbrace{2R \ln \frac{\frac{4}{3}V}{V}}_{\text{for A}} + \underbrace{R \ln \frac{\frac{2}{3}V}{V}}_{\text{for A'}} = R \ln \frac{16}{9} + R \ln \frac{2}{3} = \underline{R \ln \frac{32}{27}}$$