

HW1.

20212284 화학라 손예림.

• 계량기 용액 모델에서 두 용액의 mixing에 해당하는 Gibbs energy.

$$G_m = x_A \circ G_A + x_B \circ G_B + RT(x_A \ln x_A + x_B \ln x_B) + x_A x_B L_{A,B}$$

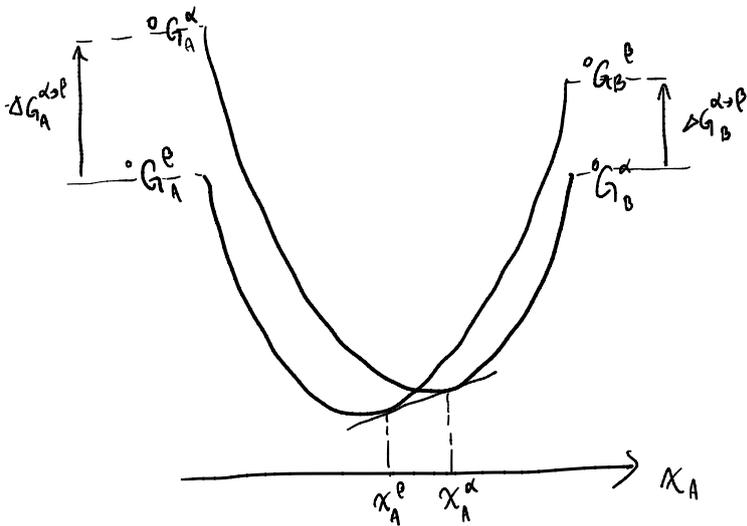
α phase 이끼 mixing에 대한 Gibbs energy.

$$G_m^\alpha = x_A (\circ G_A^{\text{ref}} + \Delta \circ G_A^{\text{ref} \rightarrow \alpha}) + x_B (\circ G_B^{\text{ref}} + \Delta \circ G_B^{\text{ref} \rightarrow \alpha}) + RT(x_A \ln x_A + x_B \ln x_B) + x_A x_B L_{A,B}^\alpha$$

β phase 이끼 mixing에 대한 Gibbs energy.

$$G_m^\beta = x_A (\circ G_A^{\text{ref}} + \Delta \circ G_A^{\text{ref} \rightarrow \beta}) + x_B (\circ G_B^{\text{ref}} + \Delta \circ G_B^{\text{ref} \rightarrow \beta}) + RT(x_A \ln x_A + x_B \ln x_B) + x_A x_B L_{A,B}^\beta$$

상평형은 두 상간의 Gibbs energy curve의 공통 접선에서 이루어진다.



$$\left. \frac{\partial G_m^\alpha}{\partial x_A} \right|_{x_A = x_A^\alpha} = \left. \frac{\partial G_m^\beta}{\partial x_A} \right|_{x_A = x_A^\beta}$$

이때  $x_A + x_B = 1$  을 이용하면 아래처럼 계산할 수 있다.

$$\begin{aligned} \frac{\partial G_m^\alpha}{\partial x_A} &= \circ G_A^{\text{ref}} + \Delta \circ G_A^{\text{ref} \rightarrow \alpha} - \circ G_B^{\text{ref}} - \Delta \circ G_B^{\text{ref} \rightarrow \alpha} + RT \left[ \ln x_A + 1 - \ln(1-x_A) - 1 \right] + (1-2x_A) L_{A,B}^\alpha \\ &= \circ G_A^{\text{ref}} + \Delta \circ G_A^{\text{ref} \rightarrow \alpha} - \circ G_B^{\text{ref}} - \Delta \circ G_B^{\text{ref} \rightarrow \alpha} + RT \ln \frac{x_A}{1-x_A} + (1-2x_A) L_{A,B}^\alpha \end{aligned}$$

$$\frac{\partial G_m^\beta}{\partial x_A} = \circ G_A^{\text{ref}} + \Delta \circ G_A^{\text{ref} \rightarrow \beta} - \circ G_B^{\text{ref}} - \Delta \circ G_B^{\text{ref} \rightarrow \beta} + RT \ln \frac{x_A}{1-x_A} + (1-2x_A) L_{A,B}^\beta$$

$$\left. \frac{\partial G_m^\alpha}{\partial x_A} \right|_{x_A=x_A^\alpha} - \left. \frac{\partial G_m^\beta}{\partial x_A} \right|_{x_A=x_A^\beta}$$

$$= \left[ \Delta G_A^{\text{ref}} + \Delta G_A^{\text{ref} \rightarrow \alpha} - G_B^{\text{ref}} - \Delta G_B^{\text{ref} \rightarrow \alpha} + RT \ln \frac{x_A^\alpha}{1-x_A^\alpha} + (1-2x_A^\alpha) L_{A,B}^\alpha \right]$$

$$- \left[ \Delta G_A^{\text{ref}} + \Delta G_A^{\text{ref} \rightarrow \beta} - G_B^{\text{ref}} - \Delta G_B^{\text{ref} \rightarrow \beta} + RT \ln \frac{x_A^\beta}{1-x_A^\beta} + (1-2x_A^\beta) L_{A,B}^\beta \right]$$

$$= \Delta G_A^{\text{ref} \rightarrow \alpha} - \Delta G_A^{\text{ref} \rightarrow \beta} - \Delta G_B^{\text{ref} \rightarrow \alpha} + \Delta G_B^{\text{ref} \rightarrow \beta} + RT \ln \frac{x_A^\alpha (1-x_A^\beta)}{x_A^\beta (1-x_A^\alpha)} + (1-2x_A^\alpha) L_{A,B}^\alpha - (1-2x_A^\beta) L_{A,B}^\beta$$

$$= -\Delta G_A^{\alpha \rightarrow \beta} + \Delta G_B^{\alpha \rightarrow \beta} + RT \ln \frac{x_A^\alpha (1-x_A^\beta)}{x_A^\beta (1-x_A^\alpha)} + (1-2x_A^\alpha) L_{A,B}^\alpha - (1-2x_A^\beta) L_{A,B}^\beta$$

$$= 0$$

$$(\because \Delta G_A^{\text{ref} \rightarrow \alpha} - \Delta G_A^{\text{ref} \rightarrow \beta} = \Delta G_A^{\text{ref} \rightarrow \alpha} + \Delta G_A^{\beta \rightarrow \text{ref}} = \Delta G_A^{\beta \rightarrow \alpha})$$

위 식에서  $x_A^\alpha, x_A^\beta$  는 reference state 이 의존 하지 않는다.

따라서 각 원소에 대해 일관된 reference state 를 사용할 때,

상평형 조성은 reference state 이 관계없이 unique 하게 결정된다.