

$$1. \Delta G = -\frac{4}{3}\pi r^3 \Delta G_v + 4\pi r^2 \gamma$$

$$nV = \frac{4}{3}\pi r^3 \Rightarrow \therefore r = \left(\frac{3nV}{4\pi}\right)^{1/3}$$

$$4\pi r^2 = 4\pi \cdot \left(\frac{3nV}{4\pi}\right)^{2/3} = (36\pi)^{1/3} n^{2/3} V^{2/3} \gamma$$

$$\therefore \Delta G = -nV \Delta G_v + (36\pi)^{1/3} n^{2/3} V^{2/3} \gamma$$

$$2. a) \Delta G = -nV \Delta G_v + (36\pi)^{1/3} n^{2/3} V^{2/3} \gamma$$

$$\therefore \frac{d\Delta G}{dn} = -V \Delta G_v + \left(\frac{32}{3}\pi\right)^{1/3} n^{-1/3} V^{2/3} \gamma$$

$$b) \frac{d\Delta G}{dn} \Big|_{n^*} = 0 = -V \Delta G_v + \left(\frac{32}{3}\pi\right)^{1/3} n^{*-1/3} V^{2/3} \gamma$$

$$n^{*-1/3} = \frac{V \Delta G_v}{\left(\frac{32}{3}\pi\right)^{1/3} V^{2/3} \gamma} \Rightarrow n^* = \frac{32\pi V^2 \gamma^3}{3 V^3 \Delta G_v^3} = \frac{32\pi \gamma^3}{3 V \Delta G_v^3}$$

$$\begin{aligned} \Delta G^* &= -n^* V \Delta G_v + (36\pi)^{1/3} n^{*2/3} V^{2/3} \gamma \\ &= -\frac{32\pi \gamma^3}{3 \Delta G_v^2} + (36\pi)^{1/3} \left(\frac{32\pi \gamma^3}{3 V \Delta G_v^3}\right)^{2/3} V^{2/3} \gamma \\ &= -\frac{32\pi \gamma^3}{3 \Delta G_v^2} + \left(36\pi \cdot \frac{1024\pi^2}{9 V^2} \cdot V^2\right)^{1/3} \cdot \frac{\gamma^2}{\Delta G_v^2} \gamma \\ &= -\frac{32\pi \gamma^3}{3 \Delta G_v^2} + \frac{16\gamma^3 \pi}{\Delta G_v^2} = \left(-\frac{32}{3} + 16\right) \cdot \frac{\pi \gamma^3}{\Delta G_v^2} \\ &= \frac{16\pi \gamma^3}{3 \Delta G_v^2} \end{aligned}$$

$$c) \Delta G_{dia} - \Delta G_{gr} = 0.$$

$$= [-n(^{\circ}G_{vap} - ^{\circ}G_{dia}) + (36\pi)^{1/3} n^{2/3} V_{dia}^{2/3} \gamma_{dia}] - [-n(^{\circ}G_{vap} - ^{\circ}G_{gr}) + (36\pi)^{1/3} n^{2/3} V_{gr}^{2/3} \gamma_{gr}]$$

$$= n(^{\circ}G_{dia} - ^{\circ}G_{gr}) + (36\pi)^{1/3} n^{2/3} (V_{dia}^{2/3} \gamma_{dia} - V_{gr}^{2/3} \gamma_{gr})$$

$$n = \left(\frac{(36\pi)^{1/3} (V_{gr}^{2/3} \gamma_{gr} - V_{dia}^{2/3} \gamma_{dia})}{^{\circ}G_{dia} - ^{\circ}G_{gr}} \right)^3$$

$$= 36\pi \left(\frac{(4 \text{ Å}^3/\text{atom} \times 3.1 \text{ J/m}^2 - 6^{2/3} \text{ Å}^3/\text{atom} \times \gamma_{dia}) (0.06242 \text{ eV/Å}^3)}{(0.02 \text{ eV/atom}) \cdot (1 \text{ J/m}^2)} \right)^3$$

$$i) \gamma_{dia} = 3.6 \text{ J/m}^2 : n \approx 464 \text{ atoms}$$

$$ii) \gamma_{dia} = 3.65 \text{ J/m}^2 : n \approx 145 \text{ atoms}$$

$$iii) \gamma_{dia} = 3.7 \text{ J/m}^2 : n \approx 21 \text{ atoms}$$

$$d) \Delta G_{dia} - \Delta G_{gr} < 0.$$

$$e) n_{gr}^* = 100 = \frac{32\pi \gamma_{gr}^3}{3 V_{gr} \Delta G_v^3} \quad \dots \text{ from 2-b)}$$

$$\therefore \Delta G_v^3 = \left(\frac{32\pi \gamma_{gr}^3}{3 V_{gr} n_{gr}^*} \right)^{1/3} = \sqrt[3]{\frac{32\pi \times 3.1 \text{ J/m}^2 \cdot (0.06242 \text{ eV/atom})}{3 \times 8 \text{ Å}^3/\text{atom} \times 100}}$$

$$= 0.067202 \text{ eV/Å}^3$$

$$f) \frac{I_{gr}}{I_{dia}} = \frac{A \cdot \exp(-\Delta G_{gr}^*/kT)}{A \cdot \exp(-\Delta G_{dia}^*/kT)} = \exp(-(\Delta G_{gr}^* - \Delta G_{dia}^*)/kT)$$

$$\Delta G_{gr}^* - \Delta G_{dia}^* = \frac{16\pi}{3} \left(\frac{\gamma_{gr}^3}{\Delta G_v^3} - \frac{\gamma_{dia}^3}{\Delta G_v^3} \right) \quad \dots \text{ from 2-b)}$$

$$V_{gr} \Delta G_v^3 - V_{dia} \Delta G_v^3 = ^{\circ}G_{dia} - ^{\circ}G_{gr} \quad \dots \text{ from 2-c)}$$

$$\Rightarrow \Delta G_v^3 = \frac{V_{gr} \Delta G_v^3 - (^{\circ}G_{dia} - ^{\circ}G_{gr})}{V_{dia}}$$

$$\therefore \frac{I_{gr}}{I_{dia}} = \exp \left(- \frac{16\pi}{3} \left(\frac{\gamma_{gr}^3}{\Delta G_v^3} - \frac{\gamma_{dia}^3}{\frac{V_{gr} \Delta G_v^3 - (^{\circ}G_{dia} - ^{\circ}G_{gr})}{V_{dia}}} \right) \right) \Bigg/ kT$$

at 1200K

$$\Delta G_v^3 = 0.067202 \text{ eV/Å}^3 \quad \dots \text{ from 2-e)}$$

$$^{\circ}G_{dia} - ^{\circ}G_{gr} = 0.02 \text{ eV/atom} \quad \dots \text{ from 2-c)}$$

$$i) \gamma_{dia} = 3.6 \text{ J/m}^2 : \frac{I_{gr}}{I_{dia}} = 2.47 \times 10^{-6}$$

$$ii) \gamma_{dia} = 3.65 \text{ J/m}^2 : \frac{I_{gr}}{I_{dia}} = 8.41 \times 10^{-2}$$

$$iii) \gamma_{dia} = 3.7 \text{ J/m}^2 : \frac{I_{gr}}{I_{dia}} = 3.83 \times 10^3$$

g) γ_{dia} 가 작아질수록 diamond가 graphite에 비해 더 안정해지며 nucleation rate가 graphite 보다 커질수 있다. 특히, 작은 양의 γ_{dia} 의 변화로도 diamond growth에 큰 영향을 끼칠수 있다.

h) $\text{CH}_4 \rightarrow \text{C} + 2\text{H}_2$ 로의 분해 반응으로 인해 반응이 활발히 일어나는 곳에서 C의 농도가 증가하여 커다란 cluster를 형성할 확률이 높아진다. 이러한 큰 cluster들은 capillarity effect를 극복하기 때문에 graphite 상태가 더 안정해진다. 따라서 높은 C의 농도가 driving force로 작용한다.