

AMSE502 Phase Transformations

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Problem Set #4

Prof. Byeong-Joo Lee
calphad@postech.ac.kr
Room 1- 311

1. Assuming a spherical nucleus, derive the following expression for the nucleation energy.

$$\Delta G = -n \nu \Delta G_s + (36\pi)^{\frac{1}{3}} n^{\frac{2}{3}} \nu^{\frac{2}{3}} \gamma$$

where n is number of atoms and ν is atomic volume.

2. (Nucleation Kinetics of CVD Diamond) Diamond is a less stable form of carbon than graphite, but it can be obtained by CVD under normal pressure and temperature.

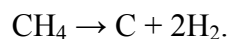
- For a spherical nucleus, express the energy change during nucleation as a function of number of atoms in cluster.
- Using the result of (a), derive the expression for the critical number of atoms and energy barrier.
- Assuming isotropic and constant surface energy for both of graphite and diamond, and using the data : $\gamma_{\text{gr}} = 3.1 \text{ Jm}^{-2}$, $\gamma_{\text{dia}} = 3.6, 3.65 \text{ and } 3.7 \text{ Jm}^{-2}$, respectively
 $v_{\text{gr}} = 8 \text{ \AA}^3/\text{atom}$, $v_{\text{dia}} = 6 \text{ \AA}^3/\text{atom}$, $^{\circ}G_{\text{dia}} - ^{\circ}G_{\text{gr}} = 0.02 \text{ eV/atom}$

For the three slightly different values of surface energy of diamond, compute the number of atoms in clusters where the stability of diamond becomes the same as that of graphite.

- What is the necessary condition for a diamond cluster of any size to be more stable than graphite ?
- Assuming that the critical number of atoms for graphite nucleation is 100, estimate the driving force for graphite nucleation.
- For the three values of surface energy of diamond, compute the ratio of nucleation rate between graphite and diamond, $I_{\text{gra}}/I_{\text{dia}}$.
For the nucleation rate, use the expression: $I = A \cdot \exp(-\Delta G^*/kT)$, and assume that A is the same constant for both of graphite and diamond.

- What is your conclusion on this problem?

- Suppose that the source of carbon in the CVD is the decomposition of CH_4 into C and H_2 .



If the C was deposited as graphite, what would be the physical meaning of driving force of graphite nucleation [the value you obtained in e)] ?